	Name
ILS	

Elements, Compounds & Mixtures Worksheet Key

Part 1: Read the following information on elements, compounds and mixtures. Fill in the blanks where necessary.

Elements:

- A pure substance containing only one kind of <u>atom</u>
- An element is always uniform all the way through (homogeneous).
- \bullet $\;$ An element $\underline{\;\;\;}$ be separated into simpler materials (except during nuclear reactions).
- Over 100 existing elements are listed and classified on the _Periodic Table_.

Compounds:

- A pure substance containing two or more kinds of <u>atoms</u>.
 The atoms are <u>chemically</u> combined in some way. Often times (but not always) they come together to form groups of atoms called molecules.
- A compound is always homogeneous (uniform).
- Compounds <u>cannot</u> be separated by physical means. Separating a compound requires a chemical reaction.
- The properties of a compound are usually different than the properties of the elements it contains.

Mixtures:

- Two or more <u>elements</u> or <u>compounds</u> NOT chemically combined.
- No reaction between substances.
- Mixtures can be uniform (called __homogeneous___) and are known as solutions.
- Mixtures can also be non-uniform (called <u>heterogeneous</u>
- Mixtures can be separated into their components by chemical or physical means.
- The properties of a mixture are similar to the properties of its components.

Part 2: Classify each of the following as elements (E), compounds (C) or Mixtures (M). Write the letter X if it is none of these.

_ E _Diamond (C)	$\underline{\mathbf{C}}$ Sugar (C ₆ H ₁₂ O ₆)	_ M _Milk	_ E _Iron (Fe)
_ M _Air	_C_Sulfuric Acid (H ₂ SO ₄)	_ M _Gasoline	_X_Electricity
_ E _Krypton (K)	_ E _Bismuth (Bi)	_ E _Uranium (U)	_M_Popcorn
$\underline{\mathbf{C}}$ _Water (H ₂ O)	_C_Alcohol (CH3OH)	_M_Pail of Garbag	e _ M _A dog
_C_Ammonia (NH:	3)_ C _Salt (NaCl)	_X_Energy	_ E _Gold (Au)
_ M _Wood	_M_Bronze	_ M _Ink	_ M _Pizza
C_Dry Ice (CO ₂)	C Baking Soda (NaHCO:	3)_ E _Titanium (Ti)	M_Concrete