



### Chemistry Unit 4 - Quiz

#### Writing Formulas & Representing Compounds

Fill in the blanks and select the ions and the compound that fits. In the space under NAME try to write a way to identify the different ions.

NAME	ION 1	ION 2	FORMULA	NAME
1. $\text{Ba}^{2+}$	iodide	$\text{I}^-$	$\text{BaI}_2$	Barium iodide
2. $\text{Na}^+$	nitrate	$\text{NO}_3^-$	$\text{NaNO}_3$	Sodium Nitrate
3. $\text{Ba}^{2+}$	oxide	$\text{O}^{2-}$	$\text{BaO}$	Barium oxide
4. $\text{K}^+$	nitrate	$(\text{NO}_3)^-$	$\text{KNO}_3$	Potassium nitrate
5. $\text{Ca}^{2+}$	oxide	$\text{O}^{2-}$	$\text{CaO}$	Calcium oxide