

Name: \_\_\_\_\_

Date: \_\_\_\_\_

# **Naming Ionic Compounds Worksheet**

$\text{Na}_2\text{CO}_3$	sodium carbonate
$\text{NaOH}$	sodium hydroxide
$\text{MgBr}_2$	magnesium bromide
$\text{KCl}$	potassium chloride
$\text{FeCl}_2$	iron (II) chloride
$\text{FeCl}_3$	iron (III) chloride
$\text{Zn(OH)}_2$	zinc hydroxide
$\text{Be}_2\text{SO}_4$	beryllium sulfate
$\text{CrF}_2$	chromium (II) fluoride
$\text{Al}_2\text{S}_3$	aluminium sulfide
$\text{PbO}$	lead (II) oxide
$\text{Li}_3\text{PO}_4$	lithium phosphate
$\text{TiI}_4$	titanium (IV) iodide
$\text{Co}_3\text{N}_2$	cobalt (II) nitride
$\text{Mg}_3\text{P}_2$	magnesium phosphide
$\text{Ga(NO}_2)_3$	gallium nitrite
$\text{Ag}_2\text{SO}_3$	silver sulfite
$\text{NH}_4\text{OH}$	ammonium hydroxide
$\text{Al(CN)}_3$	aluminium cyanide
$\text{Be(CH}_3\text{COO)}_2$	beryllium acetate