Heat and Phase Changes

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- 2. A glass of water is in the liquid phase.
- 3. An ice cube is in the Solid phase.
- 4. Solids have a definite shape and definite volume.
- 5. Describe how the molecules are arranged and moving in the liquid phase. Include a picture with your description.

particles are further apart than a gas solid and Closer than a gasallowing particles to Spread out. Energy is medium as is the intermolecular forces.



- 6. In order to get liquid water to become water vapor energy must be absorbed to the liquid water.
- 7. The phase change from liquid to gas is called vaporization. There are two types of vaporization, evaporation which occurs only at the surface while boiling occurs anywhere within liquid.
- 8. Gases have no definite Shape, and no definite volume.
- 9. Describe how the molecules are arranged and moving in the solid phase. Include a picture with your description. particles are tightly packed due to strong intermolecular forces. They only vibrate 10. Liquids have a definite Volume but not a definite shape.