

Name \_\_\_\_\_

Chemistry Worksheet  
Naming & Formula Writing (cont)

Write the formula for each of the names for the compound (underline)

- |                          |       |   |       |
|--------------------------|-------|---|-------|
| 1. Sodium nitrate        | _____ | 26. Aluminum chloride                   | _____ |
| 2. Calcium carbonate     | _____ | 27. Iron (III) hydroxide                | _____ |
| 3. Magnesium oxide       | _____ | 28. Sodium acetate                      | _____ |
| 4. Ammonium sulfide      | _____ | 29. calcium hydroxide                   | _____ |
| 5. Lead (II) sulfate     | _____ | 30. sodium iodide                       | _____ |
| 6. Sodium cyanide        | _____ | 31. Nickel (II) nitrate                 | _____ |
| 7. Potassium hydroxide   | _____ | 32. Iron (II) chloride                  | _____ |
| 8. Silver chloride       | _____ | 33. Magnesium bromide                   | _____ |
| 9. Iron (III) hydroxide  | _____ | 34. Ammonium nitrate                    | _____ |
| 10. Potassium hydroxide  | _____ | 35. Silver bromide                      | _____ |
| 11. Tin (IV) perchlorate | _____ | 36. $\text{SnCl}_4$                     | _____ |
| 12. Potassium carbonate  | _____ | 37. $\text{NH}_4$                       | _____ |
| 13. Silver nitrate       | _____ | 38. $\text{Li}_2\text{CO}_3$            | _____ |
| 14. Sodium nitrate       | _____ | 39. $\text{Cu}_2\text{O}$               | _____ |
| 15. Ammonium hydroxide   | _____ | 40. $\text{HCN}$                        | _____ |
| 16. Potassium iodide     | _____ | 41. $\text{Fe}_2\text{OCl}_2$           | _____ |
| 17. Lead (IV) oxide      | _____ | 42. $\text{BaCl}$                       | _____ |
| 18. Ammonium hydroxide   | _____ | 43. $\text{ZnSO}_4$                     | _____ |
| 19. Barium sulfate       | _____ | 44. $\text{Fe}(\text{OH})_2\text{CO}_3$ | _____ |
| 20. Barium chloride      | _____ | 45. $\text{Ca}(\text{NO}_3)_2$          | _____ |
| 21. Cobalt (II) chloride | _____ | 46. $\text{Fe}_2(\text{CO}_3)_3$        | _____ |
| 22. Sodium carbonate     | _____ | 47. $\text{NH}_4\text{Cl}$              | _____ |
| 23. Calcium oxide        | _____ | 48. $\text{CaCl}_2$                     | _____ |
| 24. Lead (II) nitrate    | _____ | 49. $\text{NaF}$                        | _____ |
| 25. Tin (IV) chloride    | _____ | 50. $\text{Cu}_2(\text{NO}_3)_2$        | _____ |