

Balancing Chemical Equations Worksheet

1. ____ H_2 + ____ $\text{O}_2 \rightarrow$ ____ H_2O
2. ____ N_2 + ____ $\text{H}_2 \rightarrow$ ____ NH_3
3. ____ S_8 + ____ $\text{O}_2 \rightarrow$ ____ SO_3
4. ____ N_2 + ____ $\text{O}_2 \rightarrow$ ____ N_2O
5. ____ $\text{HgO} \rightarrow$ ____ Hg + ____ O_2
6. ____ CO_2 + ____ $\text{H}_2\text{O} \rightarrow$ ____ $\text{C}_6\text{H}_{12}\text{O}_6$ + ____ O_2
7. ____ Zn + ____ $\text{HCl} \rightarrow$ ____ ZnCl_2 + ____ H_2
8. ____ SiCl_4 + ____ $\text{H}_2\text{O} \rightarrow$ ____ H_4SiO_4 + ____ HCl
9. ____ Na + ____ $\text{H}_2\text{O} \rightarrow$ ____ NaOH + ____ H_2
10. ____ $\text{H}_3\text{PO}_4 \rightarrow$ ____ $\text{H}_4\text{P}_2\text{O}_7$ + ____ H_2O
11. ____ $\text{C}_{10}\text{H}_{16}$ + ____ $\text{Cl}_2 \rightarrow$ ____ C + ____ HCl
12. ____ CO_2 + ____ $\text{NH}_3 \rightarrow$ ____ $\text{OC}(\text{NH}_2)_2$ + ____ H_2O
13. ____ Si_2H_3 + ____ $\text{O}_2 \rightarrow$ ____ SiO_2 + ____ H_2O_3
14. ____ $\text{Al}(\text{OH})_3$ + ____ $\text{H}_2\text{SO}_4 \rightarrow$ ____ $\text{Al}_2(\text{SO}_4)_3$ + ____ H_2O
15. ____ Fe + ____ $\text{O}_2 \rightarrow$ ____ Fe_2O_3
16. ____ $\text{Fe}_2(\text{SO}_4)_3$ + ____ $\text{KOH} \rightarrow$ ____ K_2SO_4 + ____ $\text{Fe}(\text{OH})_3$
17. ____ $\text{C}_7\text{H}_6\text{O}_2$ + ____ $\text{O}_2 \rightarrow$ ____ CO_2 + ____ H_2O
18. ____ H_2SO_4 + ____ $\text{HI} \rightarrow$ ____ H_2S + ____ I_2 + ____ H_2O
19. ____ FeS_2 + ____ $\text{O}_2 \rightarrow$ ____ Fe_2O_3 + ____ SO_2
20. ____ Al + ____ $\text{FeO} \rightarrow$ ____ Al_2O_3 + ____ Fe
21. ____ Fe_2O_3 + ____ $\text{H}_2 \rightarrow$ ____ Fe + ____ H_2O
22. ____ Na_2CO_3 + ____ $\text{HCl} \rightarrow$ ____ NaCl + ____ H_2O + ____ CO_2
23. ____ K + ____ $\text{Br}_2 \rightarrow$ ____ KBr
24. ____ C_7H_{16} + ____ $\text{O}_2 \rightarrow$ ____ CO_2 + ____ H_2O
25. ____ P_4 + ____ $\text{O}_2 \rightarrow$ ____ P_2O_5