AP Biology Biomolecules Guided Reading Worksheet 9-21-09

*Log on to the following website:

http://www.phschool.com/science/biology_place/index.html

- *click on biocoach
- *click on building biomolecules
- *click on concept 1 on the left hand navigation bar
 - 1. Give the six most abundant elements of life and their acronym.
 - **Click on the practice button and do the practice until you get all the atoms correct **Concept 2
 - Explain how these elements are joined together.
 - What determines an atoms ability to join with other atoms?
 - 4. From an electron standpoint explain what a covalent bond is.
 - 5. Explain what is meant by an atoms valence.
 - **Concept 3
 - 6. Explain why there are so many carbon compounds.
 - What is the chemistry of carbon called?
 - 8. Explain what a hydrocarbon is.
 - 9. Give 3 examples of hydrocarbons.
 - **click on the review button
 - 10. What can happen with atoms whose valence is greater than 1?
 - 11. Give a common example of a molecule formed from this process.
 - 12. What is unique about carbons 4 valence bonds?
 - **Click on the practice button and work this until you understand how to construct the model
 - **Concept 4
 - 13. Explain what a structural isomer is and how they are fomed.
 - 14. Explain what an enantiomer is and how they are formed.
 - 15. Explain what geometric isomers are and how they are formed.
 - **Concept 5
 - 16. Explain why polar molecules are formed.
 - **Click the review button
 - 17. Explain what a hydrogen bond is
 - 18. Explain how hydrogen bonds are formed in the first place.
 - 19. Give 3 examples of hydrogen bonds.
 - **Concept 6
 - 20. Explain the structure of a hydroxyl group and give an example of one.
 - 21. Explain the structure of a carbonyl group and the 2 forms they exist in.
 - 22. Explain the structure of a carboxyl group and examples of 2 molecules that form from them.
 - 23. Give the 3 types of bonds nitrogen will form.
 - 24. Explain the structure of an amino group.
 - 25. Explain the structure of the 2 molecules in the sulfhydryl group.

 - 26. Give 2 common molecules containing phosphate groups.

 **Do the self quiz until you can get 100% without using your notes.