

### **Naming Covalent Compounds(Type III) Worksheet**

Write the formulas for the following covalent compounds:

- 1) nitrogen tribromide \_\_\_\_\_
- 2) hexaboron silicide \_\_\_\_\_
- 3) chlorine dioxide \_\_\_\_\_
- 4) hydrogen iodide \_\_\_\_\_
- 5) iodine pentafluoride \_\_\_\_\_
- 6) dinitrogen trioxide \_\_\_\_\_
- 7) ammonia (nitrogen trihydride)\_\_\_\_\_
- 8) phosphorus triiodide \_\_\_\_\_
- 9) dihydrogen oxide \_\_\_\_\_
- 10) diposphorous pentoxide \_\_\_\_\_

Write the names for the following covalent compounds:

- 11) P<sub>4</sub>S<sub>5</sub> \_\_\_\_\_
- 12) O<sub>2</sub> \_\_\_\_\_
- 13) SF<sub>6</sub> \_\_\_\_\_
- 14) Si<sub>2</sub>Br<sub>6</sub> \_\_\_\_\_
- 15) SCl<sub>4</sub> \_\_\_\_\_
- 16) CH<sub>4</sub> \_\_\_\_\_
- 17) B<sub>2</sub>Si \_\_\_\_\_
- 18) NF<sub>3</sub> \_\_\_\_\_
- 19) H<sub>2</sub>O \_\_\_\_\_
- 20) N<sub>2</sub>O<sub>5</sub> \_\_\_\_\_