Name Radioactivity Worksheet	
1) 37/1 1' 4' 1 46	
a. Increases	as a chemical bond with another element its half-life
b. Decreases	
c. Remains the same	
2) A radioactive element has a half-	life of 4 hours. How much of the original radioactive sample will
remain at the end of 12 hours?	
a. 1/8	
b. 1/4	
c. 1/3	
d. 1/2	
iodine-131 that will remain will be	as of iodine 131. At the end of 32 days the number of milligrams of
a. 25.00	JC
b. 12.50	
c. 6.250	
d. 3.125	
Fill in the missing particles (X).	
4) $^{238}_{92}$ U $\rightarrow ^{234}_{90}$ Th + X	
5) $_{90}^{234}$ Th $\rightarrow _{91}^{234}$ Pa + X	
6) $_{91}^{234} \text{ Pa} \rightarrow _{-1}^{0} e + X$	
7) ${}_{4}^{9}$ Be + ${}_{2}^{4}$ He $\rightarrow {}_{6}^{12}$ C + X	
8) $_{92}^{238}$ U + $_{0}^{1}$ n \rightarrow X	
0.67: 277 2 277	
9) ${}_{3}^{6}$ Li + ${}_{1}^{2}$ H \rightarrow 2 X	
10) $_{27}^{59}$ Co + $_{2}^{4}$ He \rightarrow X	
11) 198 Au has a half life of 2.7 days. 1/8 gram in days.	A 1-gram sample of gold will have decreased to
12) What is a beta particle?	
13) What is an alpha particle?	
14) 92 U contains net	itrons.
15) $^{12}_{6}$ C contains proton	is.