

Fill in the blanks in the following table:

Nuclear Symbol	Charge	p ⁺	e ⁻	n ⁰
$^{20}_{10}\text{Ne}$	0	10	10	12
$^{81}_{35}\text{Br}$	0	35	35	46
$^{107}_{47}\text{Ag}$	0	47	47	62
$^{16}_8\text{O}$	0	8	8	10
$^{24}_{12}\text{Mg}$	0	12	12	14
$^{55}_{25}\text{Mn}$	0	25	25	31
$^{63}_{29}\text{Cu}^{+}$	1+	29	28	36
$^{32}_{16}\text{S}^{2-}$	2-	16	18	20
$^{39}_{19}\text{K}^{+}$	1+	19	18	22
$^7_3\text{Li}^{+}$	1+	3	2	4
$^{118.7}_{50}\text{Sn}^{4+}$	2+	50	48	69
$^{32}_{16}\text{S}^{2-}$	2-	16	18	17