## Chemistry: Chapter 1 **Significant Figures Worksheet**

## Count the number of sig. figs in the following measurements.

1) 67.05 mL 2) 0.003 kg 3) 45,000 L 4) 54.70 cm 5) 194.3 m<sup>2</sup> 6) 230. mm 7) 29,037 cm<sup>2</sup> 8) 0.08200 mg 9) 5ŌO g 10) 2 inches

## Solve the following arithmetic problems and give the answers with the correct number of sig. figs.

- 11) 74.862 - 3.34 12) 6.224 + 2.5392 + 4.2
- 13) 5.635 x 2 14) 23.552/3.0
- $\frac{6.93 2.0}{4.100} + \frac{1.90 \times 2.0}{1.26}$ 16) (64.78)(92) + (50.)(75.2) 15)

## Reminders of Sig. Fig. Rules

Count Rule 1:

Non-zero digits always count
In a number less than 1, leading zeroes never count
Zeroes between non-zero digits always count Count Rule 2: Count Rule 3: Zeroes at the end of decimal digits always count Count Rule 4: Count Rule 5: Zeroes in a whole number only count if there's a

decimal after them or a bar above them.

Operation Rule 1: Addition & Subtraction:

Restricted by the fewest number of decimal places

Multiplication & Division: Operation Rule 2:

Restricted by the fewest number of sig. figures