

NAME _____

Elements and Compounds 2

Directions: Determine if the examples below are Elements or Compounds based on their Chemical Formula. For Elements, use 'E'. For Compounds, use 'C'.

___1. H_2 (Triatomic Hydrogen)

___11. F_2 (Fluorine)

___2. CH_3COOH (Acetic Acid)

___12. NO_2 (Nitrogen Dioxide)



___3. Fe (Plutonium)

___13. C_{60}
(Buckminsterfullerene)

___4. COF_2 (Carbonyl fluoride)



___14. C_6H_6Fe (Ferrocene)

___5. C_4H_{10} (Butane)



___6. Cr (Chromium)

___15. $C_6H_{12}O_6$ (Glucose)

___7. $KBrO_3$ (Potassium Bromate)



___8. Mg (Magnesium)

___16. $CaCO_3$ (Chalk)

___9. H_2O (Triphosphite)



___10. C_2H_6 (Ethane)

