Worksheet Whole Chapter 11 Acids and Bases Name _____

$NaHSO_4$ sodium hydrogen sulfate

$Mg_3(PO_4)_2$ magnesium phosphate

$NaClO_3$ sodium chlorate

$NaHCO_3$ sodium hydrogen carbonate

$Ca_3(PO_4)_2$ calcium phosphate

$NaHS$ sodium hydrogen sulfide

HNO_3 nitric acid

$HBrO_3$ bromic acid

$HBrO$ hypobromous acid

H_3PO_2 hypophosphorous acid

H_2SO_4 sulfuric acid

H_2SO_3 sulfurous acid

$HClO_2$ chlorous acid

$HClO_4$ perchloric acid

$HClO$ hypochlorous acid

$HClO_3$ chloric acid

$NaClO_2$ sodium chlorite

$HClO_2$ chlorous acid

$NaClO$ sodium hypochlorite

$HClO$ hypochlorous acid

$NaHSO_4$ sodium hydrogen sulfate

$Al_2(HCO_3)_3$ aluminum hydrogen carbonate

NH_4HS ammonium hydrogen sulfide

$Cu_2(HSO_4)_2$ cupric hydrogen sulfate

H_2SO_3 sulfurous acid

$AgCl$ silver chloride

PbS lead (II) sulfide

NH_4OH ammonium hydroxide

H_2S hydrosulfuric acid, dihydrogen sulfide

HCl hydrogen chloride

$Cu(OH)_2$ cupric hydroxide

Ag_3PO_4 silver phosphate

$(NH_4)_2S$ ammonium sulfide

H_2O_2 dihydrogen peroxide

$HC_2H_3O_2$ acetic acid

$CaCO_3$ calcium (II) carbonate

HNO_2 nitrous acid

NH_3 ammonia

CO_2 carbon dioxide

$NH_4C_2H_3O_2$ ammonium acetate

$AlBr$ aluminum bromide

$SnBr_4$ tin (IV) bromide

P_2O_5 diphosphorus pentoxide

$Pb_3(PO_4)_2$ lead (II) arsenate

HCl hydrochloric acid

H_3PO_4 phosphoric acid