

Name _____

Date _____

Ionic Compounds: provide the formula

1. _____ Zinc sulfite
2. _____ Iron (III) chromate
3. _____ Magnesium hydrogen carbonate
4. _____ Iron (II) sulfate
5. _____ Copper (II) cyanide
6. _____ copper (I) chlorate
7. _____ Tin (II) nitrite
8. _____ Mercury (I) nitrate
9. _____ calcium thiosulfate
10. _____ strontium iodide
11. _____ tin (IV) chlorate
12. _____ lithium nitride
13. _____ barium peroxide
14. _____ cadmium chloride
15. _____ magnesium oxalate
16. _____ iron (II) permanganate
17. _____ copper (II) acetate
18. _____ copper (I) sulfate
19. _____ tin (II) cyanide
20. _____ mercury (I) chlorate
21. _____ calcium nitrite
22. _____ strontium nitrate
23. _____ tin (IV) thiosulfate
24. _____ lithium iodide
25. _____ barium chlorite
26. _____ cadmium nitride
27. _____ magnesium peroxide
28. _____ iron (II) chloride
29. _____ copper (II) oxalate
30. _____ cadmium permanganate
31. _____ magnesium acetate
32. _____ mercury (II) thiocyanate
33. _____ aluminum sulfide
34. _____ chromium (II) dichromate
35. _____ silver hydrogen carbonate
36. _____ ammonium oxide
37. _____ lead (II) bromide
38. _____ sodium phosphate
39. _____ lead (IV) carbonate
40. _____ potassium hydride
41. _____ zinc hydroxide
42. _____ iron (III) fluoride
43. _____ manganese (II) chromate
44. _____ mercury (II) chromate
45. _____ aluminum hydrogen sulfate
46. _____ chromium (II) thiocyanate
47. _____ silver sulfide
48. _____ ammonium dichromate
49. _____ lead (II) hydrogen carbonate
50. _____ sodium oxide
51. _____ lead (IV) bromide
52. _____ potassium phosphate
53. _____ zinc carbonate
54. _____ iron (III) hydride
55. _____ manganese (III) hydroxide
56. _____ mercury (II) fluoride
57. _____ aluminum sulfite
58. _____ lithium bromide
59. _____ ammonium dichromate
60. _____ silver nitrate