## Limiting Reagents and Percentage Yield Worksheet

1. Consider the reaction

$$I_2O_5(g) + 5 CO(g) -----> 5 CO_2(g) + I_2(g)$$

a) 80.0 grams of iodine(V) oxide, I<sub>2</sub>O<sub>5</sub>, reacts with 28.0 grams of carbon monoxide, CC

