## **Moles Worksheet**

1)	Define "mole".
2)	How many moles are present in 34 grams of Cu(OH) <sub>2</sub> ?
3)	How many moles are present in 2.45 x 10 <sup>23</sup> molecules of CH <sub>4</sub> ?
4)	How many grams are there in 3.4 x $10^{24}$ molecules of NH <sub>3</sub> ?
5)	How much does 4.2 moles of Ca(NO <sub>3</sub> ) <sub>2</sub> weigh?
6)	What is the molar mass of MgO?
7)	How are the terms "molar mass" and "atomic mass" different from one another?
8)	Which is a better unit for expressing molar mass, "amu" or "grams/mole"?
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