

# 20 Elements of Bohr Model

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## ATOMIC NOTATION

$\begin{array}{c} 1 \\ \cdot \\ \text{H} \\ 1.002 \end{array}$	<div style="border: 1px dashed black; padding: 5px; display: inline-block;"> <math>\begin{array}{c} 2 \\ \cdot \\ \cdot \\ \text{He} \\ 4.002 \end{array}</math> </div> <div style="display: inline-block; vertical-align: middle; margin-left: 10px;"> <p>← Atomic Number</p> <p>← Atomic Mass</p> </div>						$\begin{array}{c} 2 \\ \cdot \\ \cdot \\ \text{He} \\ 4.002 \end{array}$
Li	Be	B	C	N	O	F	Ne
Na	Mg	Al	Si	P	S	Cl	Ar
K	Ca	<p>Use the information from the periodic table in your book to write in the atomic number and atomic mass for each of the first 20 elements. Then use the Bohr model worksheet and your notes to draw the Lewis dot structure for each element. The first two have been completed for you as examples.</p>					