

Name \_\_\_\_\_ Date \_\_\_\_\_

## COMPOUNDS NAME AND FORMULA

Write the correct formula for the following compounds, all of which have been named using the modern Roman Numeral Method. The ionic charge is given after the first element in Roman Numerals.

cooper (I) oxide \_\_\_\_\_

mercury (I) oxide \_\_\_\_\_

gold (III) chloride \_\_\_\_\_

thallium (III) bromide \_\_\_\_\_

bismuth (V) oxide \_\_\_\_\_

terbium (III) oxide \_\_\_\_\_

uranium (VI) oxide \_\_\_\_\_

protactinium (V) bromide \_\_\_\_\_

cerium (III) oxide \_\_\_\_\_

arsenic (V) sulphide \_\_\_\_\_

tin (I) oxide \_\_\_\_\_

vanadium (II) bromide \_\_\_\_\_

magnesium (II) oxide \_\_\_\_\_

iron (II) chloride \_\_\_\_\_