

Binary Molecular Compounds Worksheet #1

Write out the name for the following compounds.

1. P_4O_{10} _____
2. $SeCl_2$ _____
3. NO _____
4. N_2O _____
5. NO_2 _____
6. SF_8 _____
7. BCl_3 _____

8. $SiCl_4$ Silicon tetrachloride

9. PCl_3 Phosphorus trichloride

10. SrF_2 Strontium fluoride

11. $Ca_3(PO_4)_2$ Calcium phosphate

12. N_2O Dinitrogen monoxide

13. NO_2 Nitrogen dioxide

14. NO Nitrogen monoxide

15. CO Carbon monoxide

16. CO_2 Carbon dioxide

17. SO_2 Sulfur dioxide

18. SO_3 Sulfur trioxide

19. $Cu(OH)_2$ Copper(II) hydroxide

20. N_2O Nitrogen monoxide

21. $Sn(NO_3)_2$ Tin(II) nitrate

22. P_2O_3 Diphosphorus trioxide

23. P_2O_5 Diphosphorus pentoxide