

**WRITE EQUATIONS**

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Date: \_\_\_\_\_

1. Zinc reacts with lead(II) nitrate to form zinc nitrate and lead.



2. Aluminium reacts with chlorine gas to form aluminium chloride.



3. Sodium phosphate reacts with calcium chloride to form calcium phosphate and sodium chloride.



4. Potassium chlorate when heated gives potassium chloride and oxygen gas.



5. Aluminium reacts with hydrochloric acid to form aluminium chloride and hydrogen gas.



6. Calcium hydroxide reacts with phosphoric acid to form calcium phosphate and water.



7. Copper reacts with sulfuric acid to form copper(II) sulfate, water and sulfur dioxide.



8. Hydrogen reacts with nitrogen to form ammonia.

