Chemical Bonding Unit - Practice Worksheets!

Review: **Ionic bonds form between _____ and non-metals. These bonds have an electronegativity difference of _____ to _____.**
Non-polar Covalent bonds form between 2 or more _____. These bonds have an electronegativity difference of _____ to _____.
Polar Covalent bonds form between two non-metals with and electronegativity difference of _____ to _____.

Directions: **Determine whether the following bonds are ionic, polar covalent or non-polar covalent. Show the dipole moment under each of the bonds.**

1. K—Br _____
2. C—O _____
3. Na—O _____
4. C—H _____
5. Br—Br _____
6. O—H _____
7. Cs—Cl _____
8. H—F _____
9. Fe—O _____
10. S—H _____