

Enzyme Virtual Lab

Go to <http://bioweb.wku.edu/courses/Biol114/enzyme/enzyme1.asp>

Answer the following Questions on paper or on this word document.

1. We start with 3 ml of H₂O₂ from your bathroom cabinet. Added Catalase. Wait three minutes. Measure and find 1 ml of H₂O₂ remaining. How much H₂O₂ was consumed in the reaction?
2. If we repeated the experiment, measured and found 2 ml of H₂O₂ remaining, would this second reaction have a higher or lower rate of reaction?
3. When you added H₂O₂, did you add Substrate or Enzyme?
4. Which reaction occurred in the tubes?
5. Which tube had the highest rate of reaction?
6. What variable caused the difference in reaction rates for each tube?
7. Why did the reaction rate decrease as the temperature changed from 34°C to 44°C?
8. Graph your results. Make sure you are graphing substrate remaining in the tubes. Provide your own scales for the graph.

Amt. of
substrate
remaining
(mL)

Temperature °C

9. **Explain** the effectiveness of catalase at different temperatures in terms of the graph. Remember that you graphed the amount of substrate left in the tube, not the amount that reacted with the catalase.