

Common polyatomic ions			
ion	name	ion	name
$\text{NH}_4^+$	ammonium	$\text{CO}_3^{2-}$	carbonate
$\text{NO}_2^-$	nitrite	$\text{HCO}_3^-$	hydrogen carbonate**
$\text{NO}_3^-$	nitrate	$\text{ClO}^-$	hypochlorite
$\text{SO}_3^{2-}$	sulfite	$\text{ClO}_2^-$	chlorite
$\text{SO}_4^{2-}$	sulfate	$\text{ClO}_3^-$	chlorate
$\text{HSO}_4^-$	hydrogen sulfate*	$\text{ClO}_4^-$	perchlorate
$\text{OH}^-$	hydroxide	$\text{C}_2\text{H}_3\text{O}_2^-$	acetate
$\text{CN}^-$	cyanide	$\text{MnO}_4^-$	permanganate
$\text{PO}_4^{3-}$	phosphate	$\text{Cr}_2\text{O}_7^{2-}$	dichromate
$\text{HPO}_4^{2-}$	hydrogen phosphate	$\text{CrO}_4^{2-}$	chromate
$\text{H}_2\text{PO}_4^-$	dihydrogen phosphate	$\text{O}_2^{2-}$	peroxide
<p>*bisulfate and **bicarbonate are widely used common names for hydrogen sulfate and hydrogen carbonate, respectively.</p>			