

Ch. 5 Worksheet

Diagonal Rule/Electron Configuration

1. Draw the diagonal rule developed by the Aufbau principle.

2. Write the electron configuration for the following elements:

a. Se #34

b. Ag #47

c. Sm #62

3. Complete the orbital notation for the following elements according to the Pauli exclusion principle and Hund's rule.

a. Be #4

1s	2s	2p	3s	3p	3d	4s
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b. P #15

1s	2s	2p	3s	3p	3d	4s
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c. Ti #22

1s	2s	2p	3s	3p	3d	4s
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4. Draw the electron dot diagram for the elements listed above.

a. Be

b. P

c. Ti

d. Se

d. Ag

e. Sm