

Periodic Table Basics

Step 1: Complete the square for each element by filling in the atomic number, name, & atomic mass.

Step 2: Determine the number of protons, neutrons, and electrons in an atom of each element.


Step 3: Identify if the element is a solid, liquid, or gas at room temperature.

Step 4: Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

Step 5: List at least three physical or chemical properties for each element.

Step 6: List at least three uses for each element.

Step 7: Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

① — B — —	P=⊕ N=— E=—	⊕ O S O L O G
M.P. = — B.P. = —		
Properties ⊕		
Uses ⊕		
Bohr Diagram ⊕		Lewis Structure
		B

Step 8: Use the following colors to shade in the square for each element. You should ONLY color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na
Orange = B & Al

Pink = O & S
Red = C & Si

Blue = Be & Mg
Tan = N & P

Purple = F & Cl
Yellow = He, Ne, & Ar

Step 9: Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

Done? Answer the questions on the worksheet using the information on your Periodic Table!



Need information? Visit the Periodic Table links on the [Chemistry](#) page of the [Kid Zone](#)!

Go to <http://sciencespot.net/> and click the Kid Zone graphic!