

**Binary Compounds: II****A. Write the chemical name for the following:  
following:**

carbon monoxide \_\_\_\_\_

carbon dioxide \_\_\_\_\_

difluorine monoxide \_\_\_\_\_

diphosphorus trioxide \_\_\_\_\_

sulphur tetrafluoride \_\_\_\_\_

ammonium chloride \_\_\_\_\_

hydrogen cyanide \_\_\_\_\_

sulphur trioxide \_\_\_\_\_

iron (II) selenide \_\_\_\_\_

gold (I) phosphide \_\_\_\_\_

tin (IV) carbide \_\_\_\_\_

manganese (II) carbide \_\_\_\_\_

**B. Write the chemical formula for the  
following:** $C_3N_4$  \_\_\_\_\_ $C_3S_2$  \_\_\_\_\_ $H_2O$  \_\_\_\_\_ $SBr_6$  \_\_\_\_\_ $CuOH$  \_\_\_\_\_ $(NH_4)_2S$  \_\_\_\_\_ $Mn_2O_7$  \_\_\_\_\_ $Al(OH)_3$  \_\_\_\_\_ $HgCN$  \_\_\_\_\_ $S_7F_2$  \_\_\_\_\_ $NO_2$  \_\_\_\_\_ $PbO_2$  \_\_\_\_\_