

CHEMISTRY-CP, PERIOD 4  
CHAPTER 11—CHEMICAL REACTIONS AND EQUATIONS  
DAILY PLANNER

DAY	CLASSWORK	HOMEWORK
1 Dec. 8	<b>Introduction to Reactions Activity (5 Pts)</b>	Finish Empirical Formula Lab
2	What is a chemical reaction? Word Equations Formula Equations	HW1: Worksheet A
3	<b>Vocabulary Quiz (20 Points)</b> Exothermic/Endothermic Balancing Equations	HW2: Worksheet B
4	Daily Quiz Synthesis/Decomposition/Combustion Reactions	HW3: Worksheet C
5	Daily Quiz Single Replacement Reactions <b>Single Replacement/Activity Series Lab (20 Points)</b>	HW4: Worksheet D
6	Daily Quiz Double-Replacement Reactions	HW5: Worksheet E
7	Daily Quiz Net Ionic Equations	HW6: Worksheet F
8	<b>Finish Daily Quiz (25 Points)</b> <b>Homework &amp; Extra Credit Due (50 Point +)</b> Chapter Review Game	Study for Chapter Test
9	Review Chapter Material	Study for Chapter Test
10 Dec. 22	<b>Chapter 11 Test (100 Points)</b> <b>Double Replacement Reactions Lab (30 Points)</b>	Finish Today's Lab

**Homework Grades:**

A(50/50) = 6/6 Homeworks Complete  
B (45/50) = 5/6 Homeworks Complete  
C (40/50) = 4/5 Homeworks Complete  
D (35/50) = 3.5/5 Homeworks Complete  
E (30/50) = 3/6 Homeworks Complete  
F (0/50) = Less than 3/6 Homeworks Complete

**Extra Credit:**

5 Points: Worksheet G--Balancing Review  
6 Points: Text p. 351 #1-12  
7 Points: Worksheet H—Midterm Review  
7 Points: Create a poster illustrating the following:  
1) The 3 things necessary for a fire to burn  
2) What the letters A, B, C, and D refer to in classifying fires  
3) The difference between "combustible" and "flammable"  
4) 2 methods of combating a fire and why it is an effective method

**Top Pair for Chapter 10**

Period 4: Hong Lee & Connor Healy

**Total Possible Points This Chapter:** 250 + 25 Possible Extra Credit Points

**Important Dates:**

Interim Progress Reports due December 16