

Chemistry Practice Sheet - Writing and Naming Formulas

1. Write the formulas of these compounds:

aluminum phosphate	_____	lithium acetate	_____
aluminum sulfide	_____	magnesium iodide	_____
aluminum sulfate	_____	magnesium perchlorate	_____
aluminum hydroxide	_____	magnesium oxalate	_____
aluminum nitrate	_____	magnesium fluoride	_____
ammonium carbonate	_____	magnesium bromide	_____
ammonium sulfide	_____	magnesium sulfate	_____
ammonium oxalate	_____	manganese (II) sulfate	_____
ammonium phosphate	_____	nickel (II) sulfide	_____
barium chromate	_____	nitrogen trioxide	_____
barium carbonate	_____	phosphorous tribromide	_____
barium hydroxide	_____	potassium carbonate	_____
barium sulfate	_____	potassium permanganate	_____
calcium fluoride	_____	potassium chlorate	_____
calcium phosphate	_____	potassium oxide	_____
calcium chromate	_____	silicon dioxide	_____
calcium acetate	_____	silver nitrate	_____
calcium perchlorate	_____	silver sulfide	_____
carbon monoxide	_____	sodium sulfate	_____
carbon tetrachloride	_____	sodium bicarbonate	_____
carbon disulfide	_____	sodium nitride	_____
chromium (II) hydroxide	_____	sodium hydrogen carbonate	_____
chromium (III) fluoride	_____	sodium fluoride	_____
cobalt (II) oxide	_____	sodium hydrogen sulfate	_____
copper (II) chloride	_____	strontium phosphide	_____
copper (I) sulfide	_____	strontium chloride	_____
copper (II) bromide	_____	strontium nitrite	_____
copper (II) oxide	_____	sulfur dioxide	_____
copper (I) oxide	_____	sulfur hexafluoride	_____
dinitrogen tetroxide	_____	tin (II) phosphate	_____
diphosphorous pentachloride	_____	tin (IV) bromide	_____
gold (III) hydroxide	_____	tin (II) fluoride	_____
iron (III) chloride	_____	tin (IV) oxide	_____
iron (III) oxide	_____	tin (IV) sulfide	_____
iron (III) hydroxide	_____	zinc nitrate	_____
iron (III) sulfide	_____	zinc sulfate	_____
iron (II) chloride	_____	zinc acetate	_____
iron (II) oxide	_____		
lead (II) carbonate	_____		
lead (IV) oxide	_____		
lithium bromide	_____		