

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

## **7-1 Practice Problems**

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Write the correct formula for each of the ionic compounds listed below.

1. Potassium Iodide

\_\_\_\_\_

2. Barium Chloride

\_\_\_\_\_

3. Lithium Bromide

\_\_\_\_\_

4. Sodium Hypochlorite

\_\_\_\_\_

5. Iron(III) Sulfate

\_\_\_\_\_

6. Chromium(III) Sulfide

\_\_\_\_\_

7. Calcium Carbonate

\_\_\_\_\_

8. Sodium Acetate

\_\_\_\_\_

9. Cobalt(II) Fluoride

\_\_\_\_\_

10. Sodium Phosphide

\_\_\_\_\_

11. Tin(IV) Oxide

\_\_\_\_\_

12. Gold(III) Bromide

\_\_\_\_\_

13. Copper(II) Iodide

\_\_\_\_\_

14. Strontium Chloride

\_\_\_\_\_

15. Lithium Acetate

\_\_\_\_\_

16. Magnesium Hydroxide

\_\_\_\_\_

17. Nickel(II) Nitrate

\_\_\_\_\_

18. Silver Oxide

\_\_\_\_\_

19. Zinc Chloride

\_\_\_\_\_

20. Magnesium Phosphate

\_\_\_\_\_