Balancing Equations Worksheet #1

Write the balanced chemical equations for the following word equations

- 1. Calcium oxide plus sulfur dioxide gives calcium sulfite
- 2. Nitrogen dioxide gas plus water yields nitric acid plus nitrogen monoxide gas
- 3. Potassium chlorate decomposes to give potassium chloride and oxygen gas
- 4. Sulfuric acid reacts with barium chloride to form barium sulfate and hydrochloric
- 5. Aluminum oxide plus sulfuric acid yields aluminum sulfate and water
- 6. Solid aluminum and lead II nitrate solution give solid lead and aluminum nitrate solution
- 7. Solid potassium chlorate decomposes to form potassium chloride and oxygen gas
- 8. Sulfuric acid decomposes to form sulfur trioxide gas and water
- 9. Propane, C_3H_8 and oxygen gas burn to form carbon dioxide gas and water
- 10. Barium oxide solid and water form barium hydroxide solution
- 11. Solid magnesium and cobalt II nitrate form magnesium nitrate and cobalt
- 12. Lithium hydroxide solution and iron III nitrate solution form iron III hydroxide solution and lithium chloride solution