

COMMON POLYATOMIC IONS:

-1	-2	-3
acetate $C_2H_3O_2^{-1}$	borate $B_4O_7^{-2}$	arsenate AsO_4^{-3}
amide NH_2^{-1}	carbonate CO_3^{-2}	citrate $C_6H_8O_7^{-3}$
astatate AtO_3^{-1}	chromate CrO_4^{-2}	ferricyanide $Fe(CN)_6^{-3}$
azide N_3^{-1}	dichromate $Cr_2O_7^{-2}$	phosphate PO_4^{-3}
bismuthate BiO_3^{-1}	fumarate $C_4H_2O_4^{-2}$	tetraborate $B_4O_7^{-3}$
bicarbonate HCO_3^{-1}	hexafluorosilicate SiF_6^{-2}	
bisulfate HSO_4^{-1}	maleate $C_4H_2O_4^{-2}$	-4
bromate BrO_3^{-1}	metasilicate SiO_3^{-2}	ferrocyanide $Fe(CN)_6^{-4}$
chlorate ClO_3^{-1}	molybdate MoO_4^{-2}	orthosilicate SiO_4^{-4}
chlorite ClO_2^{-1}	oxalate $C_2O_4^{-2}$	pyroarsenate AsO_4^{-4}
cyanate OCN^{-1}	peroxide O_2^{-2}	pyrophosphate $P_2O_7^{-4}$
cyanide CN^{-1}	peroxydisulfate $S_2O_8^{-2}$	silicate SiO_4^{-4}
formate CHO_3^{-1}	stannate SnO_3^{-2}	+1
hydroxide OH^{-1}	selenate SeO_4^{-2}	ammonium NH_4^{+1}
iodate IO_3^{-1}	succinate $C_4H_6O_4^{-2}$	phosphonium PH_4^{+1}
lactate $C_3H_5O_3^{-1}$	sulfate SO_4^{-2}	
nitrate NO_3^{-1}	tartrate $C_4H_4O_6^{-2}$	
manganate MnO_3^{-1}	tellurate TeO_4^{-2}	
metaphosphate PO_3^{-1}	thiocyanate SCN^{-2}	
propionate $C_3H_5O_3^{-1}$	thiosulfate $S_2O_3^{-2}$	
salicylate $C_7H_5O_3^{-1}$	tungstate WO_4^{-2}	
stearate $C_{18}H_{35}O_2^{-1}$		
sulfamate $NH_2SO_3^{-1}$		
thiocyanide SCN^{-1}		
vanadate VO_3^{-1}		

NOTE:		
- <i>ite</i> one less oxygen than	<i>hypo . . . ite</i> two less oxygen than	<i>per . . . ate</i> one more oxygen than
- <i>ate</i>	- <i>ate</i>	- <i>ate</i>