

Chem I

Name _____

Date _____ Per _____

Worksheet #C19: Valence Electrons and Electron Configurations and Atomic Sizes

1. Draw the arrow diagram yet again in the space below if you use it instead of the colored chart:

2. What is the full electron configuration of Cu? _____

3. Use either the arrow chart or your colored chart to write out the noble gas shortcut for these:

- | | |
|-------------|-------------|
| a. Se _____ | f. Ce _____ |
| b. Rh _____ | g. C _____ |
| c. U _____ | h. Pb _____ |
| d. Mg _____ | i. Br _____ |
| e. Sb _____ | j. Na _____ |

4. What are valence electrons? _____

5. What two kinds of orbitals contribute to the valence electrons? _____ and _____

6. How many valence electrons do each of the elements in the d and f blocks have? _____

- a. When might there be an exception to this?

7. Why are valence electrons important? _____

8. How many valence electrons do each of the following have?

- | | | | | |
|-------------|-------------|-------------|-------------|-------------|
| a. Li _____ | d. As _____ | g. Se _____ | j. Ca _____ | m. Na _____ |
| b. P _____ | e. C _____ | h. In _____ | k. Rb _____ | n. O _____ |
| c. S _____ | f. K _____ | i. Sr _____ | l. Po _____ | o. Bi _____ |