STOICHIOMETRY:	39
MOLE-MOLE PROBL	EMS

Name ____

1. $N_2 + 3H_2 \rightarrow 2NH_s$

How many moles of hydrogen are needed to completely react with two moles of nitrogen?

2. 2KCIO₃ → 2KCI + 3O₂

How many moles of oxygen are produced by the decomposition of six moles of potassium chlorate?

3. $Zn + 2HCI \rightarrow ZnCl_2 + H_2$

How many moles of hydrogen are produced from the reaction of three moles of zinc with an excess of hydrochloric acid?

4. $C_3H_8 + 5O_2 \rightarrow 3CO_2 + 4H_2O$

How many moles of oxygen are necessary to react completely with four moles of propane (C_3H_8) ?

5. $K_1PO_4 + Al(NO_3)_3 \rightarrow 3KNO_3 + AlPO_4$

How many moles of potassium nitrate are produced when two moles of potassium phosphate react with two moles of aluminum nitrate?

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