

Binary Molecular Compounds Worksheet #1

Write out the name for the following compounds.

1.  $P_4O_{10}$  \_\_\_\_\_
2.  $SeCl_2$  \_\_\_\_\_
3.  $NO$  \_\_\_\_\_
4.  $N_2O$  \_\_\_\_\_
5.  $NO_2$  \_\_\_\_\_
6.  $SF_8$  \_\_\_\_\_
7.  $BCl_3$  \_\_\_\_\_
8.  $As_2O_5$  \_\_\_\_\_
9.  $P_4S_5$  \_\_\_\_\_
10.  $Si_2Br_6$  \_\_\_\_\_

Write out the formulas for the following compounds.

11. phosphorus trichloride \_\_\_\_\_
12. arsenic pentabromide \_\_\_\_\_
13. iodine trichloride \_\_\_\_\_
14. carbon monoxide \_\_\_\_\_
15. diphosphorus pentoxide \_\_\_\_\_
16. boron trifluoride \_\_\_\_\_
17. dichlorine heptoxide \_\_\_\_\_
18. diphosphorus monosulfide \_\_\_\_\_
19. carbon tetrabromide \_\_\_\_\_
20. nitrogen trihydride \_\_\_\_\_