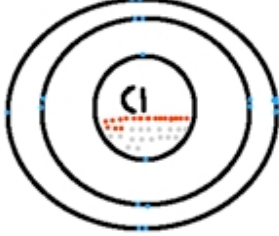

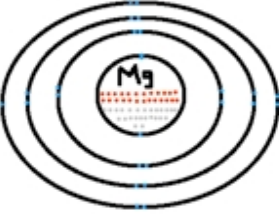





chlorine	 A Bohr model of a chlorine atom with a central nucleus labeled 'Cl' and three concentric electron shells. The outermost shell contains seven electrons, represented by blue dots.	 A Lewis dot diagram of a chlorine atom. The symbol 'Cl' is in the center, with seven blue dots representing valence electrons: two on top, two on the right, and three on the bottom.
magnesium	 A Bohr model of a magnesium atom with a central nucleus labeled 'Mg' and three concentric electron shells. The outermost shell contains two electrons, represented by blue dots.	 A Lewis dot diagram of a magnesium atom. The symbol 'Mg' is in the center, with two blue dots representing valence electrons: one on the left and one on the right.
phosphorus	 A Bohr model of a phosphorus atom with a central nucleus labeled 'P' and three concentric electron shells. The outermost shell contains five electrons, represented by blue dots.	 A Lewis dot diagram of a phosphorus atom. The symbol 'P' is in the center, with five blue dots representing valence electrons: two on top, one on the left, and two on the right.