

Name the following compounds

- 1) NaCl - Sodium Chloride
- 2) CaCO_3 - Calcium carbonate
- 3) H_2SO_4 - Sulphuric acid
- 4) CuSO_4 - Copper (II) sulphate
- 5) Na_2SO_4 - Sodium sulphate
- 6) NaOH - Sodium hydroxide
- 7) KNO_3 - Potassium nitrate
- 8) $\text{K}_2\text{Cr}_2\text{O}_7$ - Potassium dichromate
- 9) K_2MnO_4 - Potassium manganate
- 10) KMnO_4 - Potassium permanganate

Write the formulae for the following compounds

- 11) Calcium hydroxide Ca(OH)_2
- 12) Calcium nitrate $\text{Ca(NO}_3)_2$
- 13) Lead (II) oxide PbO
- 14) Lead (IV) oxide PbO_2
- 15) Magnesium nitrate $\text{Mg(NO}_3)_2$
- 16) Magnesium oxide MgO
- 17) Aluminium hydroxide Al(OH)_3
- 18) Silver nitrate AgNO_3
- 19) Lead (II) nitrate $\text{Pb(NO}_3)_2$
- 20) Lead (IV) nitrate $\text{Pb(NO}_3)_4$