

Name _____

Chemistry Worksheet
Naming & Formula Writing (cont)

Write the formula for each of the names for the compound (underline)

- | | | | |
|--------------------------|-------|---|-------|
| 1. Sodium nitrate | _____ | 26. Aluminum chloride | _____ |
| 2. Calcium carbonate | _____ | 27. Iron (III) hydroxide | _____ |
| 3. Magnesium oxide | _____ | 28. Sodium acetate | _____ |
| 4. Ammonium sulfide | _____ | 29. calcium hydroxide | _____ |
| 5. Lead (II) sulfate | _____ | 30. sodium iodide | _____ |
| 6. Sodium cyanide | _____ | 31. Nickel (II) nitrate | _____ |
| 7. Potassium hydroxide | _____ | 32. Iron (II) chloride | _____ |
| 8. Silver chloride | _____ | 33. Magnesium bromide | _____ |
| 9. Iron (III) hydroxide | _____ | 34. Ammonium nitrate | _____ |
| 10. Potassium hydroxide | _____ | 35. Silver bromide | _____ |
| 11. Tin (IV) perchlorate | _____ | 36. SnCl_4 | _____ |
| 12. Potassium carbonate | _____ | 37. NH_4 | _____ |
| 13. Silver nitrate | _____ | 38. Li_2CO_3 | _____ |
| 14. Sodium nitrate | _____ | 39. Cu_2O | _____ |
| 15. Ammonium hydroxide | _____ | 40. HCN | _____ |
| 16. Potassium nitrate | _____ | 41. Fe_2CO_3 | _____ |
| 17. Lead (IV) oxide | _____ | 42. BaO | _____ |
| 18. Ammonium hydroxide | _____ | 43. ZnO_2 | _____ |
| 19. Barium sulfate | _____ | 44. $\text{Fe}(\text{OH})_2\text{CO}_3$ | _____ |
| 20. Barium chloride | _____ | 45. $\text{Ca}(\text{NO}_3)_2$ | _____ |
| 21. Cobalt (II) chloride | _____ | 46. $\text{Fe}_2(\text{CO}_3)_3$ | _____ |
| 22. Sodium carbonate | _____ | 47. NH_4CO_3 | _____ |
| 23. Calcium oxide | _____ | 48. CaO_2 | _____ |
| 24. Lead (II) nitrate | _____ | 49. NaF | _____ |
| 25. Tin (IV) chloride | _____ | 50. $\text{Cu}_2(\text{NO}_3)_2$ | _____ |