Name	Date	Period	
Pa	eriodic Table Wor	rksheet	
Use a Periodic table to find the in			
1. What is the atomic number of:		the Atomic mass of:	
Calcium		cium	
Iron	Iron	n	
Gold	Ura	mium	
Uranium	Cop	oper	
3. How many protons do the following	have? 4. How ma	ny electrons do the following	ng have?
Calcium have		ld have	
Gold		n	
Copper		oper	
Iron	Ura	mium	
5. Does mercury have more protons and	d electrons than tin?		
6. Is mercury a heavier element than tir	1?		
7. Does potassium have more electrons	than neon?		
8. Does hydrogen have more electrons	than Uranium?		
9. Which has more protons, sulfur or ic	odine?		
10. Which has more protons, iodine or	silver?		
11. In the boxes below make Bohr moda. Determine how many electrb. Draw a Bohr model of each	ons, protons, and neutror	ns there are in each atom.	d neutrons
c. NOTE: The first energy leve			
up to 8 electrons.			
Electrons	El	lectrons	
Protons HELIUM	P	rotons OXYGEN	
N. J.		Tanahan a	
Neutrons	N	leutrons	