The	Nitrogen	Cvc	ϵ

Name

Go t	0	Nitroaer	ı Cycle	and	answer	these	questions
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- 1. What are the two conditions under which nitrogen will react with oxygen? (In other words, what is necessary for nitrogen in the air to combine with oxygen?)
- 2. What are the two compounds that are formed when nitrogen combines with oxygen?
- 3. How does nitric acid (HNO₂) form?
- 4. Why is nitric acid (HNO₂) important?
- 5. What percentage of the air we breathe is nitrogen?

Go to Nitrogen Cycle and answer these questions.

- 6. Even though considerable nitrogen is available in the air, most plants do not use the nitrogen (N₂) found in the air. Why not?
- 7. In what compounds can plants use nitrogen?
- 8. How do animals get the nitrogen they need?
- 9. Atmospheric nitrogen (N_2) is pretty <u>inert</u>. This means that it does not easily break apart. When molecules do not break apart easily, it is difficult (or impossible) for organisms to use them as a nutrient source. As a result, **nitrogen fixation** is the term used to describe the process of breaking up N_2 .
 - a. What is atmospheric fixation?
 - b. What is industrial fixation? [This is how artificial fertilizers are made.]
 - c. What is biological fixation? (In your answer, describe the types of plants associated with the symbiotic relationship.)