

A. Name the compound names.

1. P_2O_5 _____
2. K_2SO_4 _____
3. $Ca(OH)_2$ _____
4. $NaOH$ _____
5. CaO _____
6. $CaCl_2$ _____
7. $LiCl$ _____
8. CO_2 _____
9. $NaCl$ _____
10. CaO _____
11. $CaSO_4$ _____
12. Na_2O _____
13. $NaCl$ _____
14. $NaCl \cdot 2H_2O$ _____
15. H_2O _____
16. $NaOH$ _____
17. Na_2CO_3 _____
18. $NaCl$ _____
19. $NaCl$ _____
20. $CaCl_2$ _____
21. Ag_2SO_4 _____
22. Ag_2SO_4 _____
23. CO_2 _____
24. SO_2 _____
25. SO_3 _____

B. Write the formula for

1. zinc oxide _____
2. aluminum phosphate _____
3. nitrogen pentoxide _____
4. dipotassium trioxide _____
5. silver bromide _____
6. calcium hydroxide _____
7. hydrochloric acid _____
8. iron (II) hydroxide _____
9. zinc hydroxide _____
10. primary tin oxide _____
11. phosphoric acid _____
12. sodium fluoride _____
13. lithium sulfate _____
14. carbon monoxide _____
15. potassium chloride _____
16. sodium carbonate _____
17. nitrogen trihydride _____
18. sodium sulfate _____
19. phosphorus pentachloride _____
20. ammonium phosphate _____
21. copper (II) chloride _____
22. barium acetate _____
23. mercury (II) oxide _____
24. sodium chloride _____
25. dipotassium pentoxide _____