Types of Reactions Worksheet

Balance the	following	aguations	and indi	cata tha	tuno of	reaction	takina	nlaco:
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1)	NaBr + $H_3PO_4 \rightarrow$ Na $_3PO_4$ + HBr Type of reaction:
2)	Ca(OH) ₂ + Al ₂ (SO ₄) ₃ \rightarrow CaSO ₄ + Al(OH) ₃ Type of reaction:
3)	Mg + Fe₂O₃ → Fe + MgO Type of reaction:
4)	$C_2H_4 + O_2 \rightarrow CO_2 + H_2O$ Type of reaction:
5)	PbSO ₄ \Rightarrow PbSO ₃ + O ₂ Type of reaction:
6)	NH_3 + I_2 \rightarrow N_2I_6 + H_2 Type of reaction:
7)	$H_2O + M_2O + M_2SO_3 \rightarrow M_2SO_4$ Type of reaction:
8)	$H_2SO_4 +$ $NH_4OH \rightarrow$ $H_2O +$ $(NH_4)_2SO_4$ Type of reaction: